Article

Stereoselective Access to the Versatile 4-Aminohex-5-ene-1,2,3-triol Pattern

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We developed a stereocontrolled route allowing potential access to the eight isomers of 4-benzylaminohex-5-ene-1,2,3-triol in two or four steps and ca. 50% yield from readily available chiral nonracemic cis- or trans-a, β-epoxyimine precursors. A new (NH₄)₂CO₃-based carboxylation/ intramolecular cyclization sequence allowed regio- and stereocontrolled C-3 epoxide opening while neat C-2 hydrolysis was ensured by simple aqueous acidic treatment.

Introduction

Aliphatic intermediates presenting the 4-amino-1,2,3triol heading pattern have proven useful in the synthesis of various natural compounds, such as the linear amino acid portion of polyoxins¹ 1 and callipeltins² 2 (Figure 1).

This 4-amino-1,2,3-triol functional arrangement is also commonly encountered in synthetic precursors of nitrogenated heterocycles, accessible through 4- or 5-exo tet cyclization upon an activated hydroxyl group. Preparation of several azetidine.³ pyrrolidine.⁴ and indolizidines alkaloids⁵ relies, for instance, on the use of such derivatives (Figure 2).

In the course of our ongoing research program directed toward iminosugars,⁶ we developed a flexible and stereoselective approach to the 4-benzylaminohex-5-ene-1,2,3-triol intermediates in which the terminal vinyl group allows further oxidative manipulation toward various C₁- or C₂-oxygenated fragments and/or elongation by means of olefin metathesis. We wish to describe herein this work in details.

Results and Discussion

Our approach, starting from α,β -epoxyaldehydes of general structure $\mathbf{3}$, was based on three key features: (1) optical series selection during the initial Sharpless epoxidation, (2) diastereocontrolled establishment of the C-4



FIGURE 1. Amino acid moieties of polyoxines and callipeltins.



FIGURE 2. Alkaloids synthesized from a 4-amino-1,2,3-triol pattern-containing intermediate.

amino group via Felkin-Anh type organometallic reagent addition to an intermediate α,β -epoxyimine, and (3) access to the four possible diastereoisomeric arrangements through a stereochemical manifold based on the combined choice of the C-2/C-3 epoxide opening site and the cis/trans epoxide geometry (the cis series giving rise to both 2,3-syn compounds and both 2,3-anti isomers coming from the trans series).⁷

Preparation of the starting *cis*- or *trans*- α , β -epoxyaldehydes (3a or 3b) was readily accomplished from cisbut-2-ene-1,4-diol (in three steps and 78% yield or five steps and 67% yield, respectively) (not shown).8 Noteworthy is the use of o-iodoxybenzoic acid (IBX) (1.5 equiv, standard grade DMSO, rt, 5 h) to accomplish oxidation

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^a Reagents and conditions: (i) 1.0 equiv of BnNH₂, 30% w/w 4 Å MS, Et₂O, overnight, then 1.3 equiv of Et₂O·BF₃, -78 to -40 °C, 5 min, then 1.7 equiv of CH₂CHMgBr as a commercial 1.0 M solution in THF, -78°C, 2 h; *cis*-epoxide **4a**, 67% (dr >95:5); *trans*epoxide **4b** (3,4-*anti*)/**4c** (3,4-*syn*), 70% (dr 94:6); (ii) 2.7 equiv of H₂SO₄ as a 3 M aqueous solution, *p*-dioxane, reflux, 5 h; 2,3-*syn* **5a** 70% from *cis*-epoxide; 2,3-*anti* **5b** 80% from *trans*-epoxide; (iii) 8 equiv of (NH₄)₂CO₃, THF/H₂O (4:1), rt, 8-20 h, 2,3-*syn* **6a** 94% from *cis*-epoxide; 2,3-*anti* **6b** 93% from *trans*-epoxide; (iv) 1.9 equiv of TBAF on silica gel (1.25 mmol F⁻/g), THF, rt, overnight; 2,3*syn* **7a** and 2,3-*anti* **7b** 85%; (v) 6 equiv of LiOH·H₂O, *p*-dioxane/ H₂O (3:1), reflux, 8 h; 2,3-*syn* **8a** and 2,3-*anti* **8b** 95%. TBDPS = *tert*-butyldiphenylsilyl.

steps in a very convenient and efficient manner (>90% isolated yield in essentially pure material).9 Installation of the allylic amine was then inspired from Procter's Grignard reagent addition to Et₂O·BF₃-chelated in situ generated cis- α , β -epoxyimines.¹⁰ The adapted procedure involving the simple benzylamine-derived imine allowed production of the desired *cis*-epoxyamine **4a** as a single detectable isomer (67% yield from 3a) (Scheme 1). To our satisfaction, this route was extended to the trans- α , β epoxyaldehyde 3b affording the corresponding adduct 4b/ 4c with high selectivity (dr 94:6) and comparable efficiency (66% yield in major diastereoisomer 4b to which the 3,4-anti configuration was attributed by analogy with the cis series). The main byproducts isolated were the aziridines resulting from an aza-Payne type rearrangement occurring under acidic catalysis during the course of the reaction or the purification.¹¹ The anti-epoxyamine relationship, initially attributed on the basis of Procter's work, was further confirmed by means of chemical filiation and X-ray crystallography in the trans-epoxide series (vide infra).



FIGURE 3. Proposed reaction intermediate for the carboxylation/intramolecular cyclization process.

Central to our approach was the regio- and diastereocontrolled opening of the oxirane moiety. Exclusive C-2 hydrolysis was accomplished by direct aqueous acidic treatment of the C-1 silylated *cis*- and *trans*- α , β -epoxyamines **4a** and **4b**. The resulting 2,3-*syn*-3,4-*anti* and *all-anti* aminotriols **5a** and **5b** were isolated in good yields (70 and 80%, respectively). No trace of other diastereoisomers was detected in the crude reaction mixtures (¹H NMR). Such site-selective opening could arise from the deactivation of the C-3 position by the adjacent protonated amino group.¹² Aminotriol **5a** was involved in a concise total synthesis of (-)-lentiginosine and a pyrrolizidinic analogue thereof.^{6c}

Regiocontrolled C-3 opening relied on a carboxylation/ intramolecular cyclization sequence using new conditions. Treatment of epoxyamines 4a and 4b by $(NH_4)_2CO_3$ in a THF/H₂O mixture at rt smoothly delivered oxazolidinones 6a and 6b respectively in excellent yields. From a mechanistic point of view, this transformation is believed to involve attack of the secondary amine onto the CO₂ slowly released by decomposition of the carbonate salt in the aqueous medium (Figure 3). Related formation of oxazolidinones from α,β -epoxyamine under CO₂ atmosphere (MeOH, rt) has been described.¹³ The intermediate carbamic acid salt would then spontaneously cyclize into the oxazolidinone. Interestingly, the use of the ammonium cation revealed determinant for reaction progress, likely through electrophilic assistance of the oxirane opening. A study conducted with the desilylated derivative 7a (vide infra) showed that, when NaHCO₃ or K₂-CO3 were used, the presence of a catalytic amount of pyridine was necessary to observe the desired carboxylation/intramolecular cyclization process.¹⁴ It is worth emphasizing the experimental simplicity of this efficient and selective transformation.¹⁵ Moreover, the identity of the *trans*-oxazolidinone in the *cis*-epoxide series was unambiguously determined by X-ray diffraction analysis of a single crystal of *ent*-**6a** and of the corresponding benzyl ether *ent*-**6c**, both derived from α . β -epoxyaldehyde ent-3a (see the Supporting Information). These X-ray structures allowed confirmation of the previously assigned relative and absolute configuration. If one assumes that this oxazolidinone arose from an S_N2 invert-

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ing process, its trans arrangement is in agreement with the proposed anti arrangement in the starting epoxyamine. Both oxazolidinones **6a** and **6b**, presenting three differentiated alcohol functions, constitute an interesting platform for further synthetic developments.

The targeted aminotriols were finally prepared from intermediates **6a** and **6b** by hydroxyl group desilylation, delivering primary alcohols **7a** and **7b** in 85% yield, followed by cyclic carbamate basic hydrolysis (95% yield). That way, the *all-syn* derivative **8a** and the 2,3-*anti*-3,4*anti* isomer **8b** were readily obtained from the corresponding oxazolidinones in good overall yield. Access to derivative **6a** already allowed a total synthesis of 1,4dideoxy-1,4-imino-glucitol.^{6a}

Conclusion

In summary, we developed a stereocontrolled route offering potential access to the eight isomers of the 4-benzylaminohex-5-ene-1,2,3-triol in two or four steps (ca. 50% yield in both cases) from readily available chiral nonracemic *cis*- or *trans*- α , β -epoxyimine precursors. Note-worthy is the use of a practical and efficient carboxylation/intramolecular cyclization sequence involving the trivial (NH₄)₂CO₃ salt as reagent. Synthetic work further exemplifying the versatility of these highly substituted synthetic intermediates will be reported in due course.

Experimental Section

Compound 4a. Benzylamine (687 µL, 5.88 mmol) was added dropwise to a solution of α,β -cis-epoxyaldehyde **3a** (2.00 g, 5.88 mmol) in freshly distilled Et₂O (40 mL) containing powder activated 4 Å molecular sieves (30% w/w). The resulting mixture was vigorously stirred at room temperature until no starting material could be detected by IR analysis (ca. 3 h, disappearance of C=O bond 1729 cm⁻¹ and appearance of C= N bond 1663 cm⁻¹). The solution was then cooled to -78 °C, and freshly distilled BF3*Et2O (950 µL, 7.64 mmol) was added dropwise at this temperature. The mixture was allowed to stir at $-40~^\circ\mathrm{C}$ for 5 min before being cooled again to $-78~^\circ\mathrm{C}.$ Vinylmagnesium bromide in THF (10.00 mL of a 1 M commercial solution, 10.00 mmol) was then added dropwise over 5 min, and the mixture was vigorously stirred at -78 °C for 2 h. The reaction was quenched by portionwise addition of a THF/saturated aqueous NH₄Cl mixture (20:80, 20 mL) over 1 h 30 at -78 °C with vigorous stirring before being allowed to warm to room temperature. The mixture was then filtered through Celite, and the solvents were evaporated off under reduced pressure. The resulting suspension was extracted with Et_2O (3 × 100 mL), and the combined extracts were successively washed with water and brine, dried over Na_2SO_4 , filtered, and concentrated under reduced pressure. The pale yellow oil thus obtained was first filtered through a plug of silica gel eluting with petroleum ether/EtOAc (90:10) to remove magnesium salt. Concentration of the filtrate to dryness under reduce pressure gave a crude product that was then purified by medium-pressure column chromatography on silica gel treated with 2.5% v/v Et₃N eluting with petroleum ether/ EtOAc (gradient 95:5 to 80:20) to afford epoxyamine 4a (1.8 g, 3.94 mmol, 67% yield): $R_f = 0.21$ (petroleum ether/EtOAc 87:12 under a saturated atmosphere of NH₄OH); $[\alpha]^{25}_{D}$ +6.5 (c 1.23, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.72-7.62 (m, 4H, Ph), 7.50-7.35 (m, 6H, Ph), 7.30-7.20 (m, 5H, Ph), 5.88 (ddd, ${}^{3}J_{\text{H5H4}} = 7.5 \text{ Hz}$, ${}^{3}J_{\text{H5H6}} = 10.3 \text{ Hz}$, and ${}^{3}J_{\text{H5H6}} = 17.4 \text{ Hz}$, 1H, H-5), 5.31 (brd, ${}^{3}J_{\text{H6H5}} = 10.3 \text{ Hz}$, 1H, H-6), 5.27 (brd, ${}^{3}J_{\text{H6H5}} = 17.2 \text{ Hz}$, 1H, H-6'), 3.83–3.80 (m, 2H, 2 × H-1), 3.76 (AB of an ABX, ${}^{2}J_{gem} = 13.4$ Hz, $\Delta\delta a - \delta b = 62.9$ Hz, 2H, NCH₂-Ph), 3.24-3.19 (m, 1H, H-4), 3.07 (dd, ${}^{3}J_{H3H4} = 3.93$ Hz and ${}^3J_{\rm H3H2}=8.0$ Hz, 1H, H-3), 3.00 (pseudot, ${}^3J_{\rm H2H1}={}^3J_{\rm H2H3}=8.0$ Hz, 1H, H-2), 1.07 (s, 9H, C(CH_3)_3); ${}^{13}{\rm C}$ NMR (100 MHz, CDCl_3) δ 140.00 (Cquat arom), 137.6 (C-5), 135.7, 135.6 (CH arom), 133.3, 133.1 (Cquat arom), 129.9, 128.4, 128.1, 127.9, 127.8, 127.0 (CH arom), 118.2 (C-6), 62.0 (C-1), 59.0 (C-4), 58.8 (C-3), 56.9 (C-2), 50.9 (NCH_2Ph), 26.9 (CH_3), 19.2 (C(CH_3)_3); IR (thin film) 3154 (N-H), 1633 (C=C), 1108 (C-O) cm^{-1}; MS (DCI/NH_3) m/z 458 (MH^+, 100); HRMS (DCI/NH_3) m/z calcd for C_{29}H_{36}NO_2Si 458.2515, found 458.2512.

Compound 4b. trans- α , β -Epoxyamine 4b (2.22 g, 4.85 mmol, 66% yield) was prepared from α,β -trans-epoxyaldehyde 3b (2.5 g, 7.35 mmol) using the procedure described for the preparation of compound **4a**: $R_f = 0.24$ (petroleum ether/ EtOAc 90:10 under a saturated atmosphere of NH_4OH ; [α]²⁵_D $-0.6 (c = 1.94, \text{ CHCl}_3); {}^{1}\text{H} \text{ NMR} (400 \text{ MHz}, \text{CDCl}_3) \delta 7.80 -$ 7.70 (m, 5H, Ph), 7.50–7.28 (m, 10H, Ph), 5.81 (ddd, ${}^{3}J_{H5H4} =$ 7.8 Hz, ${}^{3}J_{\rm H5H6'}$ = 10.2 Hz, and ${}^{3}J_{\rm H5H6}$ = 17.3 Hz, 1H, H-5), 5.36 (brd, ${}^{3}J_{\text{H6H5}} = 17.3$ Hz, 1H, H-6), 5.34 (brd, ${}^{3}J_{\text{H6'H5}} = 10.2$ Hz, 1H, H-6'), 3.85 (AB_q, ${}^{2}J_{gem} = 13.3$ Hz, 2H, NCH₂Ph, $\Delta\delta a - \delta b$ = 57.0 Hz), 3.83 (AB of ABX, ${}^{3}J_{H1H2} = 3.5$ Hz, ${}^{3}J_{H1'H2} = 4.6$ Hz, and ${}^{2}J_{gem} = 11.8$ Hz, 2H, 2 × H-1, $\Delta \delta a \cdot \delta b = 33.0$ Hz), 3.30 (dd, ${}^{3}J_{H4H3} = 4.8 \text{ Hz and } {}^{3}J_{H4H5} = 7.8 \text{ Hz}$, 1H, H-4), 3.26–3.22 (m, 1H, H-2), 3.04 (dd, ${}^{3}J_{H3H2} = 2.2 \text{ Hz and } {}^{3}J_{H3H4} = 4.8 \text{ Hz}$, 1H, H-3), 1.85-1.77 (m, 1H, N-H), 1.14 (s, 9H, C(CH₃)₃); ¹³C NMR (100 MHz, CDCl₃) δ 140.4 (Cquat arom), 137.0 (C-5), 135.8 (CH arom), 133.5 (Cquat arom), 130.0, 128.7, 128.4, 128.0, 127.2 (CH arom), 118.7 (C-6), 64.0 (C-1), 60.7 (C-4), 58.0 (C-3), 56.4 (C-2), 51.3 (NCH₂Ph), 27.1 (CH₃), 19.5 (C(CH₃)₃); IR (thin film) 2931 (N–H), 1588 (C=C), 1110 (C–O) cm⁻¹; MS (DCI/NH₃) m/z 458 (MH⁺, 100); HRMS (DCI/NH₃) m/z calcd for C₂₉H₃₆NO₂Si 458.2515, found 458.2514.

Compound 4c. In the course of the preparation of **4b** the 3,4-syn minor diastereoisomer ${\bf 4c}$ was also isolated in 4%yield: $R_f = 0.22$ (petroleum ether/EtOAc 90:10 under a saturated atmosphere of NH₄OH); $[\alpha]^{25}$ _D -0.6 (c = 1.94, CHCl₃); ¹H NMR (400 MHz, CDCl₃) & 7.80-7.65 (m, 5H, Ph), 7.30-6.80 (m, 10H, Ph), 5.89-5.79 (m, 1H, H-5), 5,34-5,28 (m, 2H, 2 × H-6), 3.86 (AB_q, ${}^{2}J_{gem} = 13.3$ Hz, 2H, NCH₂Ph) $\Delta\delta a - \delta b = 47.9 \text{ Hz}$, 3.82 (AB of ABX, ${}^{3}J_{\text{H1H2}} = 3.4 \text{ Hz}$, ${}^{3}J_{\text{H1'H2}}$ = 4.6 Hz, $^2\!J_{gem}$ = 11.9 Hz, 2H, 2 \times H-1) $\Delta\delta a - \delta b$ = 38.5 Hz, 3.14 (m, 1H, H-2), 3.04 - 3.00 (m, 2H, H-3, H-4), 2.10 - 1.85 (m, 2H, H-3, H-4), 2.10 - 1.85 (m, 2H, H-3)1H, N-H), 1.10 (s, 9H, C(CH₃)₃); ¹³C NMR (100 MHz, CDCl₃) δ 140.3 (Cquat arom), 136.5 (C-5), 135.8 (CH arom), 133.5 (Cquat arom), 130.0, 128.7, 128.4, 128.0, 127.2 (CH arom), 118.3 (C-6), 63.1 (C-1), 62.6 (C-4), 58.8 (C-3), 56.7 (C-2), 51.3 (NCH₂-Ph), 27.0 (CH₃), 19.5 (C(CH₃)₃); IR (thin film) 2931 (N-H), 1660 (C=C), 1110 (C-O) cm⁻¹; MS (DCI/NH₃) m/z 458 (MH⁺) 100); HRMS (DCI/NH₃) m/z calcd for C₂₉H₃₆NO₂ Si 458.2515, found 458.2512.

Compound 5a. A 3 M aqueous H₂SO₄ solution (2.91 mL, 8.72 mmol) was added to a solution of $cis-\alpha,\beta$ -epoxyamine 4a (500 mg, 1.09 mmol) in *p*-dioxane (10 mL). The mixture was refluxed for 5 h and allowed to cool before neutralization with a 3 M aqueous NaOH solution (6 mL, 18.00 mmol) followed by solid NaHCO₃. p-Dioxane was then evaporated off under reduced pressure and the resulting aqueous phase extracted with CH_2Cl_2 (3 × 100 mL) and with EtOAc (2 × 50 mL). The combined organic phases were successively washed with water and brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude product was purified by flash column chromatography on deactivated silica gel (treated with 2.5% v/v Et₃N) eluting with EtOAc/Et₂O/MeOH (gradient 85: 10:5 to 75:10:15) to afford aminotriol **5a** (180 mg, 0.76 mmol, 70% yield): $R_f = 0.17$ (EtOAc/Et₂O/MeOH, 80:10:10 under a saturated atmosphere of NH₄OH); $[\alpha]^{25}_{D}$ +1.0 (*c* 6.00, CHCl₃); ¹H NMR (250 MHz, CDCl₃) δ 7.38–7.25 (m, 5H, Ph), 5.87– 5.77 (m, 1H, H-5), 5.36 (brd, ${}^{3}\!J_{\rm H6H5} =$ 11.9 Hz, 1H, H-6), 5.32 (brd, ${}^{3}J_{\rm H6'H5} = 18.1$ Hz, 1H, H-6'), 3.85-3.82 (m, 1H, H-1), 3.81-3.78 (m, 1H, H-1'), 3.76 (m, 1H, H-2), 3.75-3.72 (m, 1H, H-3), 3.73 (AB_q, ${}^{2}J_{gem} = 13.0$ Hz, 2H, NCH₂Ph, $\Delta \delta a - \delta b = 112.2$ Hz), 3.49 (brdd, ${}^{3}J_{H4H3} = 4.8$ Hz and ${}^{3}J_{H4H5} = 8.4$ Hz, 1H, H-4); $^{13}\mathrm{C}$ NMR (63 MHz, CDCl_3) δ 139.5 (Cquat arom), 136.4 (C-5),

128.7, 128.5, 127.4 (CH arom), 119.8 (C-6), 75.1 (C-3), 70.8 (C-2), 65.4 (C-1), 64.3 (C-4), 51.2 (NCH₂Ph); IR (thin film) 3421 (O-H), 1636 (C=C), 1060 (C-O) cm⁻¹; MS (DCI/NH₃) *m/z* 238 (MH⁺, 100); HRMS (DCI/NH₃) *m/z* calcd for $C_{13}H_{20}NO_3$ 238.1443, found 238.1446.

Compound 5b. By applying to *trans*- α , β -epoxyamine **4b** (400 mg, 0.87 mmol) the procedure described for the preparation of compound 5a, aminotriol 5b was obtained (165 mg, 0.70 mmol, 80% yield): $R_f = 0.18$ (EtOAc/Et₂O/MeOH, 80:10:10 under a saturated atmosphere of NH₄OH); $[\alpha]^{25}D - 11.9$ (c 0.80, CH₃OH); ¹H NMR (400 MHz, CD₃OD /D₂O) δ 7.40-7.20 (m, 5H, Ph), 5.80 (ddd, ${}^{3}J_{H5H4} = 8.9$ Hz, ${}^{3}J_{H5H6} = 10.3$ Hz, and $_{3J_{\rm H5H6'}}$ = 17.2 Hz, 1H, H-5), 5.38 (dd, $^{2}J_{gem}$ = 1.9 Hz and $^{3}J_{\rm H6H5}$ 33 10 112 112 112 112 1133 11333 11333 11333 11333 11333 11333 11333 113 Hz and ${}^{2}J_{gem} = 11.4$ Hz, 1H, H-1), 3.61 (dd, ${}^{3}J_{H3H2} = 3.0$ Hz and ${}^{3}J_{H3H4} = 5.0$ Hz, 1H, H-3), 3.60 (dd, ${}^{3}J_{H1'H2} = 5.2$ Hz and ${}^{2}J_{gem} = 11.4$ Hz, 1H, H-1′), 3.57 - 3.51 (m, 1H, H-2), 3.38 (brdd, ${}^{3}J_{\rm H4H3}^{'} = 5.0 \text{ Hz and } {}^{3}J_{\rm H4H5} = 8.9 \text{ Hz}, 1\text{H}, \text{H-4}); {}^{13}\text{C} \text{ NMR} (100)$ MHz, CD₃OD /D₂O) δ 139.1 (Cquat arom), 135.5 (C-5), 128.5, 128.4, 127.1 (CH arom), 118.9 (Ĉ-6), 73.5 (Ĉ-3), 72.3 (Ĉ-2), 63.6 (C-4), 63.4 (C-1), 50.3 (NCH₂Ph); IR (thin film) 3365 (O-H), 1661 (C=C), 1025 (C-O); MS (DCI/NH₃) *m/z* 238 (MH⁺, 100); HRMS (DCI/NH₃) m/z calcd for C₁₃H₁₉NO₃ 238.1443, found 238.1443

Compound 6a. To a solution of $cis - \alpha, \beta$ -epoxyamine **4a** (400 mg, 0.87 mmol) in THF/water 4:1 (20 mL) was added $(NH_4)_2$ - CO_3 (670 mg, 7.00 mmol), and the heterogeneous mixture was vigorously stirred at room temperature until TLC analysis showed no remaining starting material (ca. 20 h). The THF was then evaporated off under reduced pressure and the resulting aqueous phase extracted with Et_2O (3 × 50 mL). The combined organic phases were successively washed with water and brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude product was purified by flash column chromatography on silica gel eluting with petroleum ether/EtOAc (80:20) to afford oxazolidinone 6a (410 mg, 0.82 mmol, 94% yield): $R_f = 0.25$ (petroleum ether/EtOAc 80:20); $[\alpha]_{25}$ D +30.2 (c 1.06, CHCl₃); ¹H NMR (250 MHz, CDCl₃) δ 7.63-7.59 (m, 5H, Ph), 7.43-7.26 (m, 10H, Ph), 5.67 (ddd, ${}_{3J_{H5H4}} = 8.9$ Hz, ${}^{3}J_{H5H6} = 9.9$ Hz, and ${}^{3}J_{H5H6'} = 16.8$ Hz, 1H, H-5), 5.33 (d, ${}^{3}J_{H6H5} = 9.3$ Hz, 1H, H-6), 5.22 (d, ${}^{3}J_{H6'H5} = 16.8$ Hz, 1H, H-6'), 4.39 (AB_q, ${}^{2}J_{gem} = 15.1$ Hz, 2H, NCH₂Ph, Δδa-δb = 181.0 Hz), 4.25 (dd, ${}^{3}J_{H3H2} = 2.3$ Hz and ${}^{3}J_{H3H4} = 7.1$ Hz, 1H, H-3), 4.05 (dd, ${}^{3}J_{H4H3} = 6.9$ Hz and ${}^{3}J_{H4H5} = 9.1$ Hz, 1H, H-4), 3.82–3.55 (m, 3H, 2 \times H-1 and H-2), 2.39–2.21 (m, 1H, O–H), 1.04 (s, 9H, C(CH_3)_3); ^{13}C NMR (63 MHz, CDCl_3) δ 157.5 (C=O), 135.6 (Cquat arom), 135.6 (C-5), 135.5 (CH arom), 134.6 (Cquat arom), 132.9, 132.8 (Cquat arom), 130.0, 128.7, 128.3, 127.9 (CH arom), 122.9 (C-6), 78.0 (C-3), 70.8 (C-2), 64.1 (C-1), 60.0 (C-4), 46.0 (NCH₂Ph), 27.0 (CH₃), 19.2 (C(CH₃)₃); IR (thin film) 3405 (O-H), 1726 (C=O), 1635 (C=C), 1064 (C-O) cm⁻¹; MS (DCI/NH₃) m/z 519 (MNH₄⁺, 100); HRMS (DCI/NH₃) m/z calcd for C₃₀H₃₆NO₄Si 502.2414, found 502.2415.

Compound 6b. Oxazolidinone 6b (508 mg, 1.01 mmol, 93% yield) was prepared from $trans-\alpha,\beta$ -epoxyamine **4b** (500 mg, 1.09 mmol) using the procedure described for the preparation of compound **6a**: $R_f = 0.23$ (petroleum ether/EtOAc 88:12); $[\alpha]_{25}$ D +38.8 (c 1.19, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.70-7.60 (m, 5H, Ph), 7.50–7.20 (m, 10H, Ph), 5.74 (ddd, ${}^{3}J_{H5H4} =$ $8.8 \text{ Hz}, {}^{3}J_{\text{H5H6}} = 10.0 \text{ Hz}, \text{ and } {}^{3}J_{\text{H5H6}'} = 17.1 \text{ Hz}, 1\text{H}, \text{H-5}), 5.34$ $(1H, d, {}^{3}J_{H6H5} = 10.0 \text{ Hz}, \text{H-6}), 5.24 (d, {}^{3}J_{H6'H5} = 17.0 \text{ Hz}, 1H,$ H-6'), 4.41 (AB_q, ${}^{2}J_{gem} = 15.1$ Hz, 2H, NCH₂Ph, $\Delta\delta a - \delta b = 314.8$ Hz), 4.27 (pseudot, ${}^{3}J_{\rm H3H2} = {}^{3}J_{\rm H3H4} = 5.8$ Hz, 1H, H-3), $4.12 \text{ (dd, } {}^{3}J_{H4H3} = 5.7 \text{ Hz and } {}^{3}J_{H4H5} = 8.6 \text{ Hz}, 1H, H-4), 3.85-$ 3.69 (m, 3H, 2 × H-1 and H-2), 2.71 (d, ${}^{3}J_{OH-H2} = 4.7$ Hz, 1H, O-H), 1.08 (s, 9H, C(CH₃)₃); 13 C NMR (100 MHz, CDCl₃) δ 157.5 (C=O), 135.9 (Cquat arom), 135.7 (CH arom), 135.3 (C-5), 132.9, 132.8 (Cquat arom), 130.2, 128.9, 128.5, 128.2, 128.1, 128.0 (CH arom), 121.2 (C-6), 78.3 (C-3), 71.9 (C-2), 63.6 (C-1), 59.7 (C-4), 45.9 (NCH₂Ph), 27.1 (CH₃), 19.5 (C(CH₃)₃); IR (thin film) 3421 (O–H), 1741 (C=O), 1638 (C=C), 1109 (C–O) cm⁻¹; MS (DCI/NH₃) *m/z* 519 (MNH₄⁺, 100); HRMS (DCI/NH₃) *m/z* calcd for $C_{30}H_{37}NO_4Si$ 502.2414, found 502.2411

Compound 7a. TBAF on silica gel (0.89 g at ca. 1.25 mole of fluoride/g, ca. 0.71 mmol) was added to a solution of silyl ether 6a (300 mg, 0.59 mmol) in anhydrous THF (20 mL). The reaction mixture was vigorously stirred until TLC analysis showed no remaining starting material (ca. 20 h) and then filtered. Silica gel was rinsed several times with ethyl acetate $(3 \times 70 \text{ mL})$, and the combined filtrates were concentrated in vacuo. The resulting crude product was purified by flash column chromatography on silica gel eluting with CH₂Cl₂/ MeOH (gradient 100:0 to 95:5) to afford primary alcohol 7a (133 mg, 0.51 mmol, 85% yield): $R_f = 0.20$ (CH₂Cl₂/MeOH 95: 5); $[\alpha]^{25} {}_{\rm D}$ +71.6 (c 0.55, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.27–7.18 (m, 5H, Ph), 5.61 (ddd, ${}^{3}J_{H5H4} = 8.9$ Hz, ${}^{3}J_{H5H6} =$ 9.9 Hz, and ${}^{3}J_{\rm H5H6'} = 16.9$ Hz, 1H, H-5), 5.27 (d, ${}^{3}J_{\rm H6H5} = 10.0$ Hz, 1H, H-6), 5.20 (d, ${}^{3}J_{\text{H6'H5}} = 17.0$ Hz, 1H, H-6'), 4.32 (AB_q, ${}^{2}J_{gem} = 15.3 \text{ Hz}, 2\text{H}, \text{NC}H_{2}\text{Ph}, \Delta\delta a - \delta b = 245.5 \text{ Hz}), 4.17 \text{ (dd,}$ ${}^{3}J_{\text{H3H2}} = 2.6 \text{ Hz and } {}^{3}J_{\text{H3H4}} = 7.1 \text{ Hz}, 1\text{H}, \text{H-3}), 4.05 \text{ (dd, } {}^{3}J_{\text{H4H3}} = 7.1 \text{ Hz and } {}^{3}J_{\text{H4H5}} = 8.9 \text{ Hz}, 1\text{H}, \text{H-4}), 3.69-3.56 \text{ (m, 4H, 2)}$ \times H-1, H-2 and O-H); ¹³C NMR (100 MHz, CDCl₃) δ 158.1 (C=O), 135.6 (Cquat arom), 134.5 (C-5), 128.8, 128.3, 127.9 (CH arom), 122.2 (C-6), 79.4 (C-3), 71.1 (C-2), 63.2 (C-1), 60.5 (C-4), 46.1 (NCH₂Ph); IR (thin film) 3346 (O-H), 1729 (C=O), 1440 (C=C), 1079 (C-O) cm⁻¹; MS (DCI/NH₃) m/z 281 (MNH4⁺, 100); HRMS (DCI/NH3) m/z calcd for C14H18NO4 264.1236, found 264.1235

Compound 7b. By applying to silyl ether 6b (452 mg, 0.90 mmol) the procedure described for the preparation of compound 7a, alcohol 7b (210 mg, 0.80 mmol, 85% yield) was obtained: $R_f = 0.22$ (CH₂Cl₂/MeOH 95:5); $[\alpha]^{25}$ D +82.0 (c 1.22, CHCl₃); ¹H NMR (400 MHz, CDCl₃/D₂O) & 7.40-7.20 (m, 5H, Ph), 5.69 (ddd, ${}^{3}J_{H5H4} = 8.8$ Hz, ${}^{3}J_{H5H6} = 10.0$ Hz, and ${}^{3}J_{H5H6'} =$ 17.0 Hz, 1H, H-5), 5.32 (d, ${}^{3}J_{H6H5} = 10.0$ Hz, 1H, H-6), 5.26 (d, ${}^{3}J_{\text{H6'H5}} = 17.0 \text{ Hz}, 1\text{H}, \text{H-6'}), 4.36 (AB_{q}, {}^{2}J_{gem} = 15.2 \text{ Hz}, 2\text{H},$ NCH₂Ph, $\Delta \delta a - \delta b = 290.1$ Hz), 4.19 (pseudot, ${}^{3}J_{H3H2} = {}^{3}J_{H3H4}$ = 5.6 Hz, 1H, H-3), 4.11 (dd, ${}^{3}J_{H4H3}$ = 5.9 Hz and ${}^{3}J_{H4H5}$ = 8.6 Hz, 1H, H-4), 3.84–3.78 (m, 1H, H-2), 3.63–3.52 (m, 2H, 2 × H-1); ¹³C NMR (100 MHz, CDCl₃/D₂O) δ 158.0 (C=O), 135.7 (Cquat arom), 135.0 (C-5), 129.0, 128.4, 128.1 (CH arom), 121.6 (C-6), 80.0 (C-3), 71.9 (C-2), 62.2 (C-1), 59.7 (C-4), 46.0 (NCH₂-Ph); IR (thin film) 3429 (O-H), 1723 (C=O), 1634 (C=C), 1060 (C-O) cm⁻¹; MS (DCI/NH₃) m/z 281 (MNH₄⁺, 100); HRMS (DCI/NH₃) m/z calcd for C₁₄H₁₈NO₄ 264.1236, found 264.1239.

Compound 8a. LiOH·H₂O (485 mg, 11.4 mmol) was added to a solution of oxazolidinone 7a (500 mg, 1.90 mmol) in p-dioxane/water 3:1 (20 mL). The mixture was refluxed until TLC showed no remaining starting material (ca. 8 h) and allowed to cool before neutralization by solid NaHCO₃. p-Dioxane was then evaporated off under reduced pressure and the resulting aqueous phase extracted with CH_2Cl_2 (3 \times 70 mL) and with EtOAc (2 \times 50 mL). The combined organic phases were successively washed with water and brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude product was purified by flash column chromatography on deactivated silica gel (treated with 2.5% v/v Et_3N) eluting with EtOAc/ Et_2O /MeOH (90:10:0 to 75:10:15) to afford aminotriol 8a (428 mg, 1.81 mmol, 95% yield): $R_f =$ 0.22 (EtOAc/Et₂O/MeOH, 85:10:5 under a saturated atmosphere of NH₄OH); $[\alpha]^{25}$ _D -20.0 (*c* 2.75, CHCl₃); ¹H NMR (400 \dot{M} Hz, CDCl₃/D₂O) δ 7.40–7.24 (m, 5H, Ph), 5.79 (ddd, ${}^{3}J_{H5H4}$ = 8.5 Hz, ${}^{3}\!J_{\rm H5H6}$ = 10.3 Hz and ${}^{3}\!J_{\rm H5H6'}$ = 17.2 Hz, 1H, H-5), 5.35 (ddd, ${}^{4}J_{H6H4} = 0.5$ Hz, ${}^{2}J_{gem} = 1.5$ Hz and ${}^{3}J_{H6H5} = 10.3$ Hz, 1H, H-6), 5.24 (ddd, ${}^{4}J_{H6'H4} = 0.8$ Hz, ${}^{2}J_{gem} = 1.5$ Hz, and ${}^{3}J_{H6'H5} = 17.2$ Hz,1H, H-6'), 3.75 (AB_q, ${}^{2}J_{gem} = 12.8$ Hz, 2H, NCH₂Ph, $\Delta\delta a - \delta b = 107.7$ Hz), 3.75–3.73 (m, 2H, 2 × H-1), $3.72 \text{ (dd, } {}^{3}J_{\text{H2H3}} = 1.3 \text{ Hz and } {}^{3}J_{\text{H2H1}} = 3.3 \text{ Hz}, 1\text{H}, \text{H-2}$, 3.62 $(dd, {}^{3}J_{H3H2} = 1.3 \text{ Hz and } {}^{3}J_{H3H4} = 4.7 \text{ Hz}, 1H, H-3), 3.22 (dd, 3.2)$ ${}^{3}J_{\rm H4H3} = 4.7$ Hz and ${}^{3}J_{\rm H4H5} = 8.5$ Hz, 1H, H-4); ${}^{13}C$ NMR (100 MHz, CDCl₃/D₂O) & 139.2 (Cquat arom), 136.3 (C-5), 128.8, 128.6, 127.6 (CH arom), 118.9 (C-6), 74.1 (C-3), 72.4 (C-2), 65.4

(C-1), 63.7 (C-4), 50.5 (NCH₂Ph); IR (thin film) 3453 (O–H), 1638 (C=C), 1066 (C–O) cm⁻¹; MS (DCI/NH₃) m/z 238 (MH⁺, 100); HRMS (DCI/NH₃) m/z calcd for C₁₃H₁₉NO₃ 238.1443, found 238.1449.

Compound 8b. Aminotriol **8b** (214 mg, 0.90 mmol, 95% yield) was prepared from oxazolidinone **7b** (250 mg, 0.95 mmol) using the procedure described for the preparation of compound **8a**: $R_f = 0.22$ (EtOAc/Et₂O/MeOH 80:10:10 under a saturated atmosphere of NH₄OH); $[\alpha]^{25}_{D} - 5.8$ (c 1.38, CH₃OH); ¹H NMR (400 MHz, CDCl₃/D₂O) δ 7.35–7.15 (m, 5H, Ph), 5.75 (ddd, ${}^{3}J_{H5H4} = 8.7$ Hz, ${}^{3}J_{H5H6} = 10.2$ Hz, and ${}^{3}J_{H6H5} = 17.2$ Hz, 1H, H-6), 5.18 (d, ${}^{3}J_{H6H5} = 17.2$ Hz, 1H, H-6'), 3.65 (AB_q, ${}^{2}J_{gem} = 12.7$ Hz, 2H, NCH₂Ph, $\Delta \delta a - \delta b = 108.9$ Hz), 3.68–3.50 (m, 4H, 2 × H-1, H-2 and H-3), 3.15 (dd, ${}^{3}J_{H4H3} = 4.9$ Hz and ${}^{3}J_{H4H5} = 8.7$ Hz, 1H, H-4); ¹³C NMR (100 MHz, CDCl₃/D₂O) δ 139.1 (Cquat arom), 136.4 (C-5), 128.7, 128.6, 127.5 (CH arom), 119.2

(C-6), 73.7 (C-3), 73.5 (C-2), 62.8 (C-4), 62.7 (C-1), 50.6 (NCH₂-Ph); IR (thin film) 3426 (O-H), 2920 (N-H), 1637 (C=C), 1061 (C-O); (DCI/NH₃) m/z 238 (MH⁺, 100); HRMS (DCI/NH₃) m/z calcd for $C_{13}H_{20}NO_3$ 238.1443, found 238.1444.

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Supporting Information Available: General experimental methods. X-ray structure of the *trans*-oxazolidinone *ent*-**6a**. CIFs for compounds *ent*-**6a** and *ent*-**6c**. This material is available free of charge via the Internet at http://pubs.acs.org.

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